







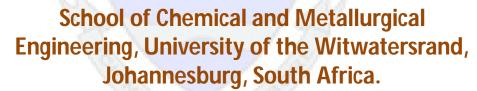






Mathematical Modelling, Design and Optimization of Symmetric Electrochemical **Supercapacitors from Layered Materials** OF THE WITH PARENTAL **Presented By** 

Innocent S. Ike, Iakovos Sigalas, Sunny Iyuke and Kenneth I. Ozoemena



**Department of Chemical Engineering, Federal** University of Technology, Owerri, Nigeria.



Mathematical modelling, design, and optimization of electrochemical capacitors from layered materials lke, I. S.; lakovos, S.; lyuke, S. and Ozoemena, K. I. is licensed under a Creative Commons Attribution-NonCommercial-NoDerivatives 4.0 International License.



### PRESENTATION OUTLINE

- Objectives
- Overview
- Mathematical Models
- Results
- Discussion of Results
- Conclusions

### **OBJECTIVES**

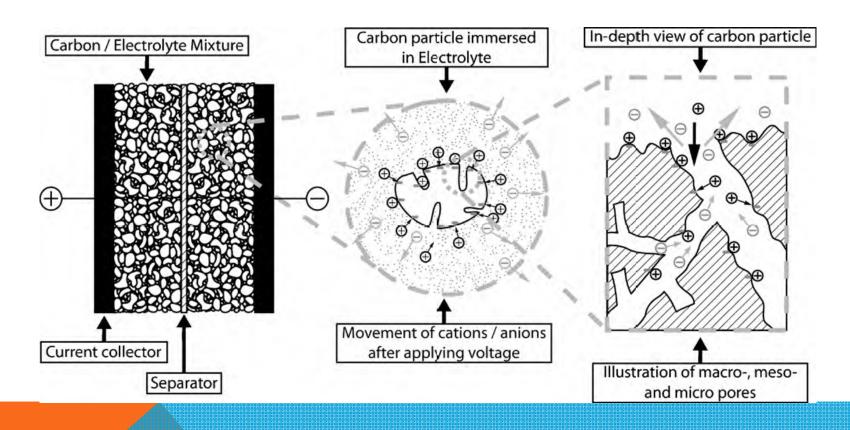
- ➤ To develop a mathematical description of of ESs subject to characteristic parameters of cell layers;
- ➤ To build theoretical basis for calculations and improvement of energy and other parameters of ESs;
- To provide ideal alternative to time-consuming numerous experiments for design and optimization of ESs cell;
- To optimize specific energy, power density, etc, of ESs subject to characteristic properties of its

layers;

To develop an Optimal Process Design of the Loss

Cell.

# ILLUSTRATION OF THE BASIC COMPONENTS AND DESIGN OF AN EDLS



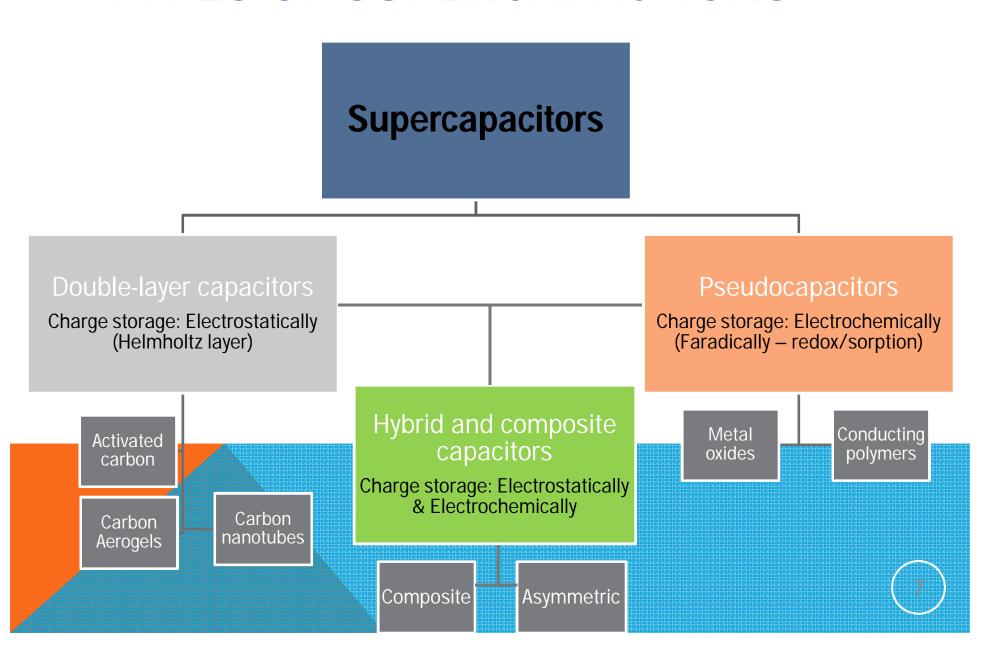
# Advantages

- Long life: It works for large number of cycle without wear and aging.
- Rapid charging: it takes seconds to charge completely
- High power storage: It stores huge amount of energy in a small volume.
- Faster release: Release the energy much faster than battery.
- > Low toxicity of materials used.
- > High cycle efficiency: 95% or more (low ESR).

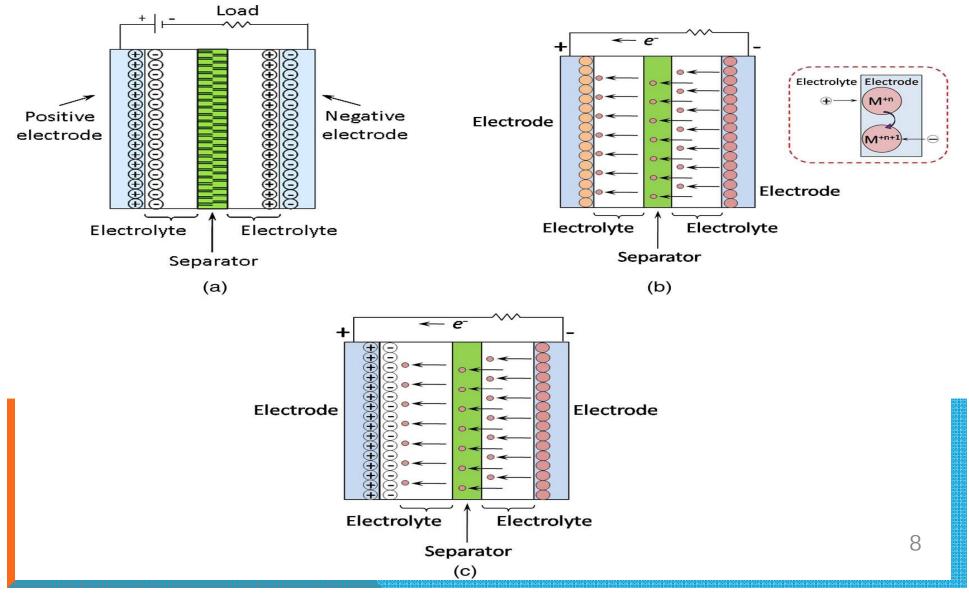
# Disadvantages

- They have low energy density
- Individual cell shows low voltage
- They have high self-discharge as compared to battery.
- Voltage balancing is required when more than three capacitors are connected in series.
- \* The voltage varies with the energy stored.
- To effectively store and recover energy requires sophisticated electronic control and switching equipment.

### TYPES OF SUPERCAPACITORS



# SCHEMATIC REPRESENTATION OF SUPERCAPACITOR TYPES: (A) EDLS TYPE; (B) PSEUDOCAPACITORS TYPE; (C) HYBRID CAPACITOR TYPE (THE FIGURE IS NOT TO SCALE)



# HOMOGENEOUS (SYMMETRIC) SUPERCAPACITORS WITH ONLY EDLS ELECTRODES.

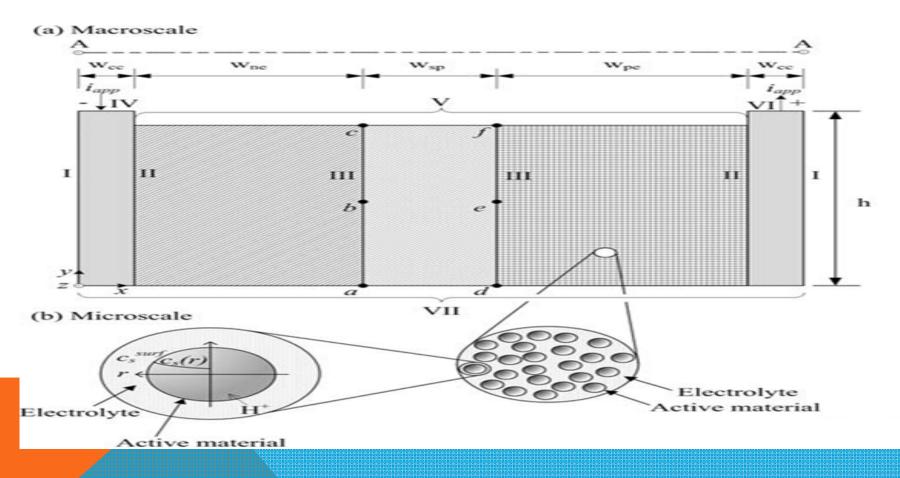
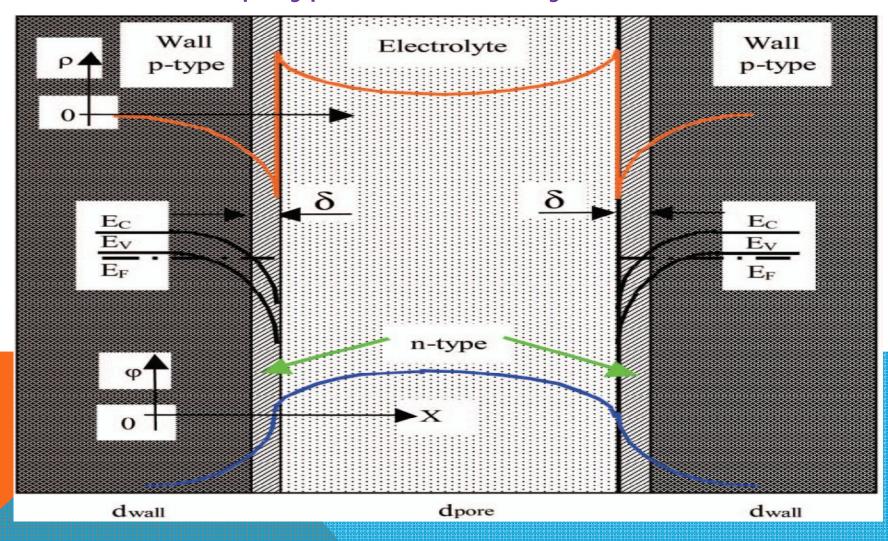


Figure 1. A symmetric supercapacitors cell with only EDL electrodes showing (a) various functional layers on the macroscale, and (b) the active material on the electrodes on the microscale.

Distributions of electric charge and potentials in the walls of cylindrical shaped pores and in the electrolyte which is inside the pores of the negative electrode with p-type conductivity.



#### THE MODEL EQUATIONS WITHOUT SELF-DISCHARGES TERM

$$\frac{\partial \varphi_{ne}(x,t)}{\partial t} = \beta^2 \frac{\partial^2 \varphi_{ne}(x,t)}{\partial x^2}$$

$$\beta^2 = \frac{\alpha_1 \alpha_2}{C_V (\alpha_2 + \alpha_1)}$$

$$J_{VR}(x,t) = J_{VR1}(x,t) = J_{VR2}(x,t)$$

The boundary conditions of this task during the capacitor discharge are as follows:

$$J_1(0,t) = -\alpha_1 \frac{\partial \varphi_{ne}(x,t)}{\partial x}\Big|_{x=0} = J_0$$

$$J_{2}(w,t) = \alpha_{2} \frac{\partial \varphi_{ne}(x,t)}{\partial t} \Big|_{x=w_{ne}} = J_{0}$$

and the initial condition is as follows:

$$\varphi_{ne}(x,t)|_{t=0} = \varphi_{ne}^{0-}$$

The boundary conditions of this task during the capacitor charge are as follows:

$$J_{1}(0,t) = \alpha_{1} \frac{\partial \varphi_{ne}(x,t)}{\partial x} \Big|_{x=0} = J_{0}$$

$$J_{2}(w,t) = -\alpha_{2} \frac{\partial \varphi_{ne}(x,t)}{\partial x} \Big|_{x=w_{ne}} = J_{0}$$

and the initial condition is as follows:

$$\varphi_{ne}(x,t)|_{t=0} = \varphi_{ne}^{0^+}$$

## Solutions of the Supercapacitors Models without Self-discharges

Solving equation 1 with the boundary and initial conditions, we obtain equation 10 below:

$$\varphi_{ne}(x,t) = \varphi_{ne}^{0^{+}} \pm \frac{J_{0}t}{w_{ne}C_{V}} \pm \frac{J_{0}C_{V}}{w_{ne}C_{V}} \left[ \frac{(\alpha_{2} + \alpha_{1})}{2\alpha_{1}\alpha_{2}} x^{2} - \frac{w_{ne}x}{\alpha_{2}} + \frac{w_{ne}^{2}(2\alpha_{2} - \alpha_{1})}{6\alpha_{1}\alpha_{2}} - \frac{2w_{ne}^{2}}{\pi^{2}n^{2}} \right]$$

$$\sum_{n=1}^{\infty} \left( \frac{1}{\alpha_1} + \frac{(-1)^n}{\alpha_2} \right) \ell^{\frac{\beta^2 n^2 \pi^2 t}{w_{ne}^2}} Cos \left( \frac{n\pi x}{w_{ne}} \right)$$

where the sign

$$\frac{J_0}{w_{ne}C_V}$$
 < 0 (sign "—") represents the process of the supercapacitors charge  $\frac{J_0}{w_{ne}C_V}$  > 0 (sign "+") represents the process of the supercapacitors discharge.

# VOLTAGE OF THE SUPERCAPACITORS WITHOUT SELF-DISCHARGE TERM

The voltage V(x,t) of the supercapacitors as a function of its thickness and time of charge and discharge is given as:

$$V(x,t) = -\frac{1}{2}\varphi_{ne}(w_{ne},t) - \frac{1}{2}\varphi_{ne}(0,t)$$

On substituting  $x = w_{ne}$  and x = 0 into equation 10, equation 11 will be as follow:

$$V(w_{ne},t) = V_{0^{\pm}} \pm \frac{J_0}{w_{ne}C_V} \left\{ t + \frac{C_V w_{ne}^2(\alpha_2 + \alpha_1)}{2\alpha_2 \alpha_1} X \left[ \frac{1}{6} - \frac{1}{\pi^2} \sum_{n=1}^{\infty} \frac{1}{n^2} \ell^{-\frac{2n\pi}{w_{ne}}^2 \left[ \frac{\alpha_2 \alpha_1 t}{C_V(\alpha_2 + \alpha_1)} \right]} \right] \right\}$$
 12

Where 
$$V_{0^+}=-\varphi_{ne}^{0^+}$$
 and the sign  $\frac{J_0}{w_{ne}C_{ne}}<0$  (sign "–") represents the process of the supercapacitors discharge

$$\frac{J_0}{w_{ne}C_v}>0$$
 (sign "+") represents the process of the supercapacitors charge.

When the effective conductivities of the electrode and the electrolyte are the same, that is,  $\alpha_1 = \alpha_2 = \alpha$  equation 12 has the following expression

$$V(w_{ne},t) = V_{0^{\pm}} \pm \frac{J_0}{w_{ne}C_V} \left\{ t + \frac{C_V w_{ne}^2}{\alpha} X \left[ \frac{1}{6} - \frac{1}{\pi^2} \sum_{n=1}^{\infty} \frac{1}{n^2} e^{-\left(\frac{2n\pi}{w_{ne}}\right)^2 \left[\frac{\alpha t}{2C_V}\right]} \right] \right\}$$
 13

Electromotive force (emf) of supercapacitors is determined as

$$U_{S}(w_{ne}, w_{sp}, w_{pe}, \beta, C_{V}, t) = \varphi_{pe}(0, t) - \varphi_{ne}(0, t)$$
14

From Equation 31 we obtain an analytical expression for emf of the supercapacitors during its charge and discharge by constant current as:

$$U_{S}(d,t) = U_{S0} \pm \frac{J_{0}}{C_{V}} w_{S1} \left\{ t + C_{V} w_{S2} \left[ \frac{(2\alpha_{2} - \alpha_{1})}{6\alpha_{1}\alpha_{2}} - \frac{2}{\pi^{2}} w_{S2} \sum_{n=1}^{\infty} \frac{1}{n^{2}} \ell^{-\left(\frac{2n\pi}{w_{S1}}\right)^{2} \left[\frac{\alpha_{2} \alpha_{1} t}{C_{V}(\alpha_{2} + \alpha_{1})}\right]} \right\} 15$$

where  $U_{S0}=arphi_{pe}^{0^+}-arphi_{ne}^{0^+}$  is the emf of the supercapacitors at t=0;

$$w_{S1} = \left(\frac{1}{w_{pe}} - \frac{1}{w_{ne}}\right)$$
 and  $w_{S2} = (w_{pe} - w_{ne})$ 

#### **VOLTAGE OF THE SUPERCAPACITORS WITHOUT SELF-**DISCHARGE TERM CONT.

Because supercapacitors has specific internal resistance,  $\rho_{int} \Omega c m^2$ , the supercapacitors voltage during the charge and discharge is expressed by the formula:

$$U(w_{e},t) = U_{0} \pm J_{0} \rho_{\text{int}} \pm \frac{J_{0}w_{S1}}{C_{V}} \left\{ t + C_{V}w_{S2} \left[ \frac{(2\alpha_{2} - \alpha_{1})}{6\alpha_{1}\alpha_{2}} - \frac{2w_{S2}}{\pi^{2}} \sum_{n=1}^{\infty} \frac{1}{n^{2}} \ell^{-\frac{(2n\pi)^{2}}{w_{S1}}} \right] \right\} 16$$

Recall that the effective conductivities of the electrode  $\mathfrak{g}_1$ and the electrolytes  $\alpha_2$  are respectively defined by equations 17 and 18

below:  

$$\alpha_1 = \sigma_{ne}^s + 2D_e C_V$$
  
 $\alpha_2 = \sigma_{ne}^l + 2D_+ C_V$ 

$$=\sigma_{ne}^l + 2D_+C_V$$

### **RESULTS**

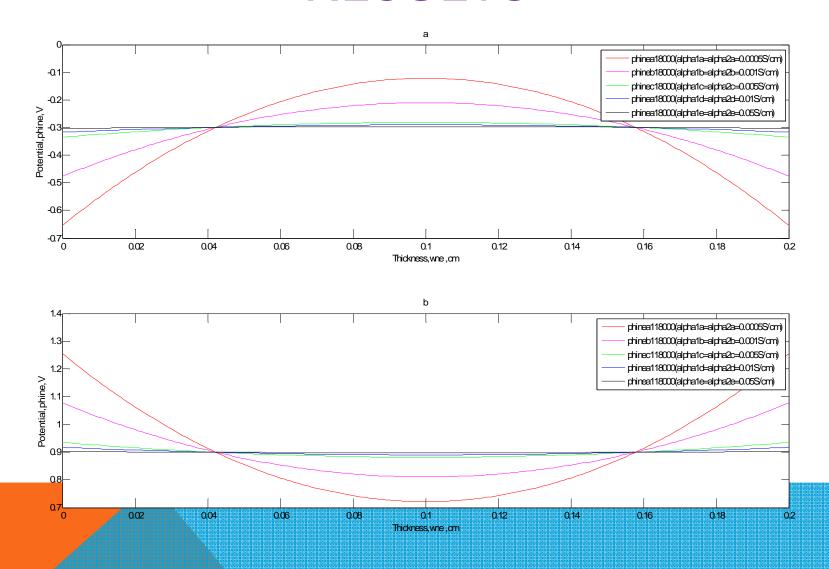


Figure 2. Distribution of potential of the negative electrode as a function of position after (a) 5 h charge and (b) 5 h discharge by constant current of supercapacitors with different parameters of  $\alpha_1$  and  $\alpha_2$ 

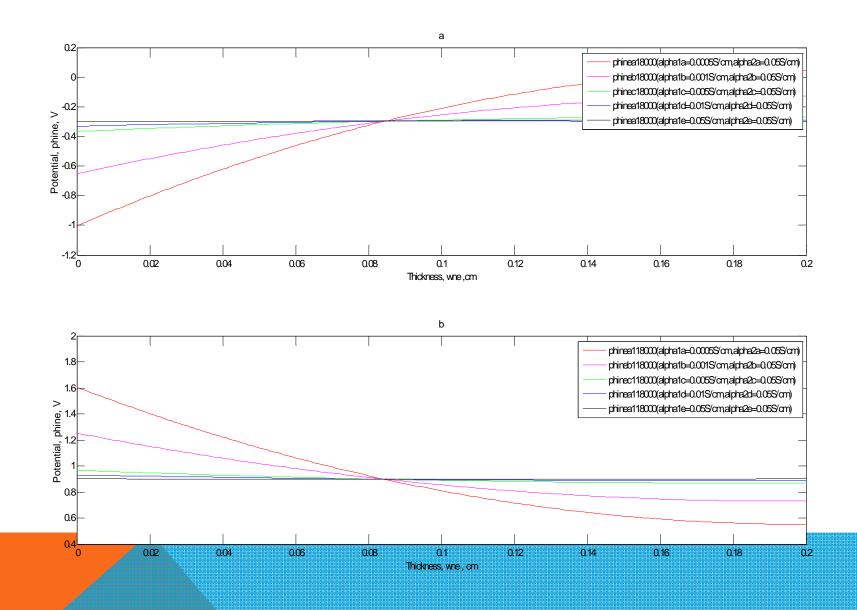


Figure 3. Distribution of potential of the negative electrode as a function of position after (a) 5 h charge and (b) 5 h discharge by constant current of supercapacitors with different parameters of  $\alpha_1$ 

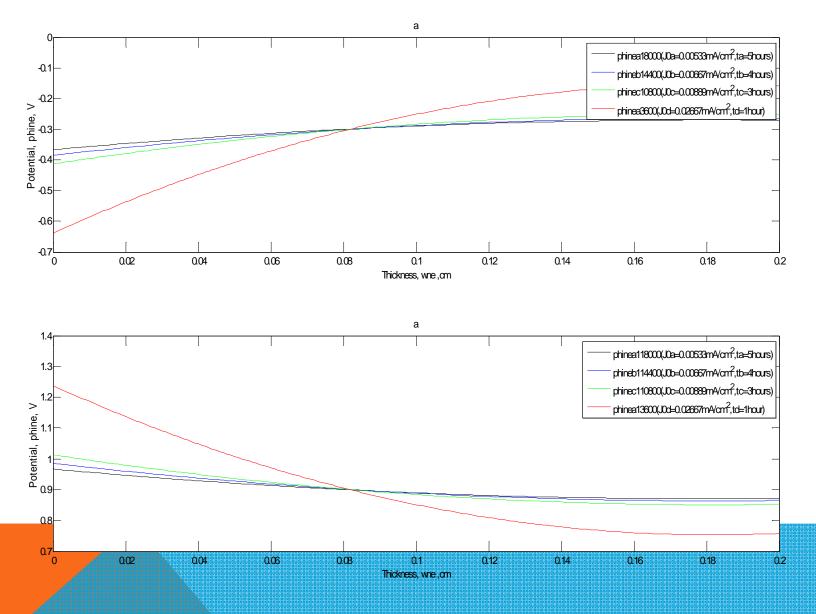


Figure 4. Distribution of potential of megative electrode as a function of position during (a) charge and (b) discharge of supercapacitors by constant current of different charge and discharge time.

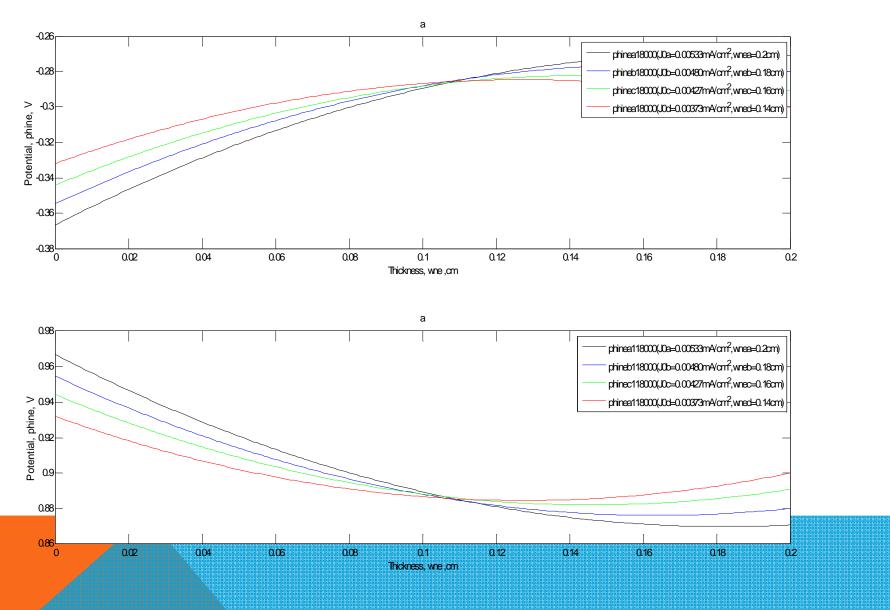


Figure 5. Distribution of potential of negative electrodes as a function of position at (a) 5 h charge and (b) 5 h discharge by constant current of supercapacitors with different thickness of negative electrode.

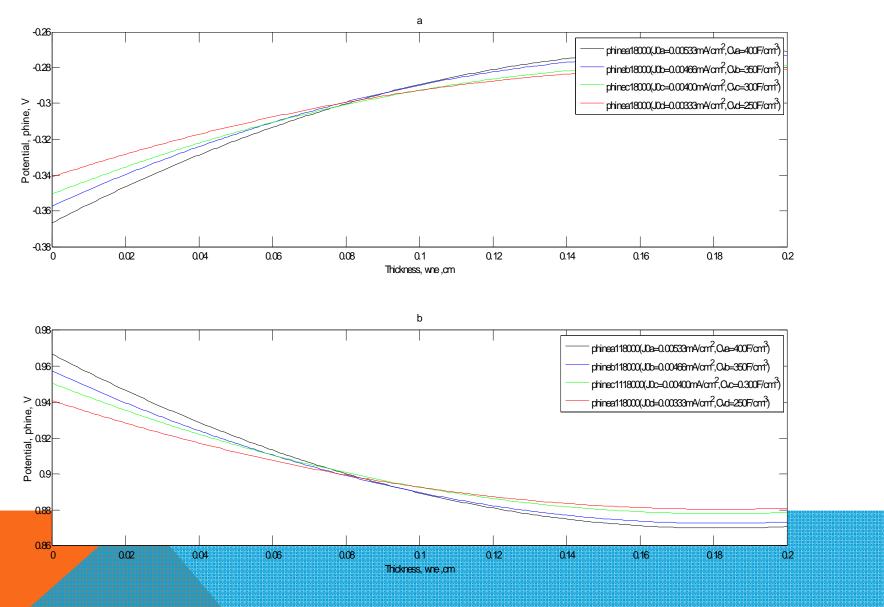
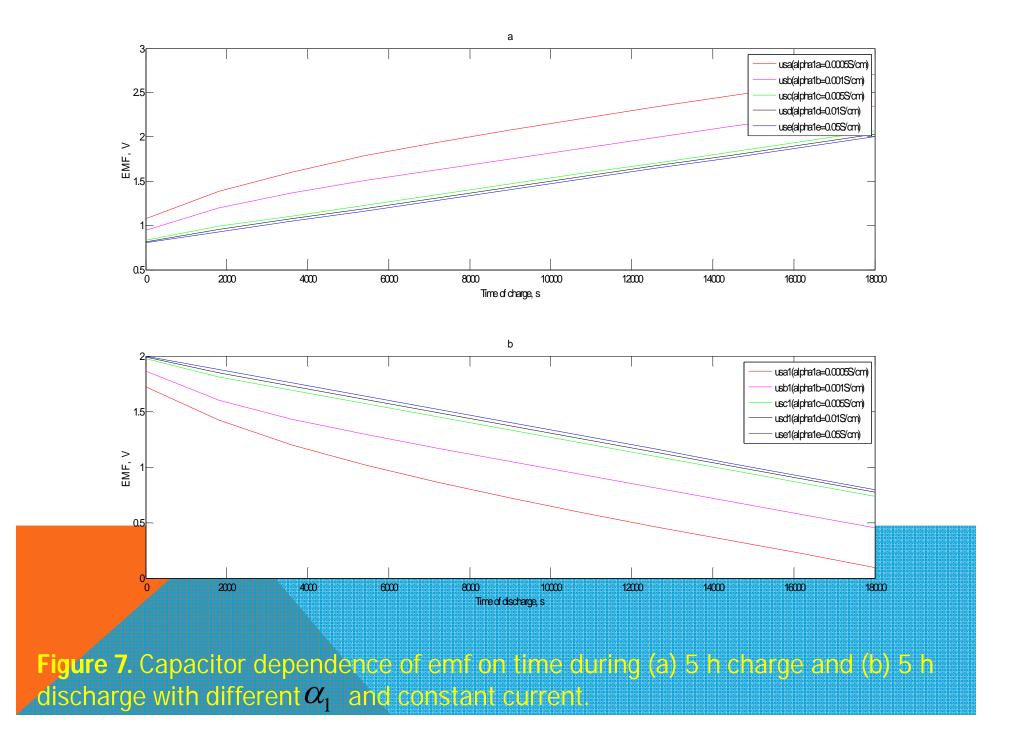


Figure 6. Distribution of potential of negative electrode as a function of position at (a) 5 h charge and (b) 5 h discharge of supercapacitors with different values of specific capacitance by constant current of different densities



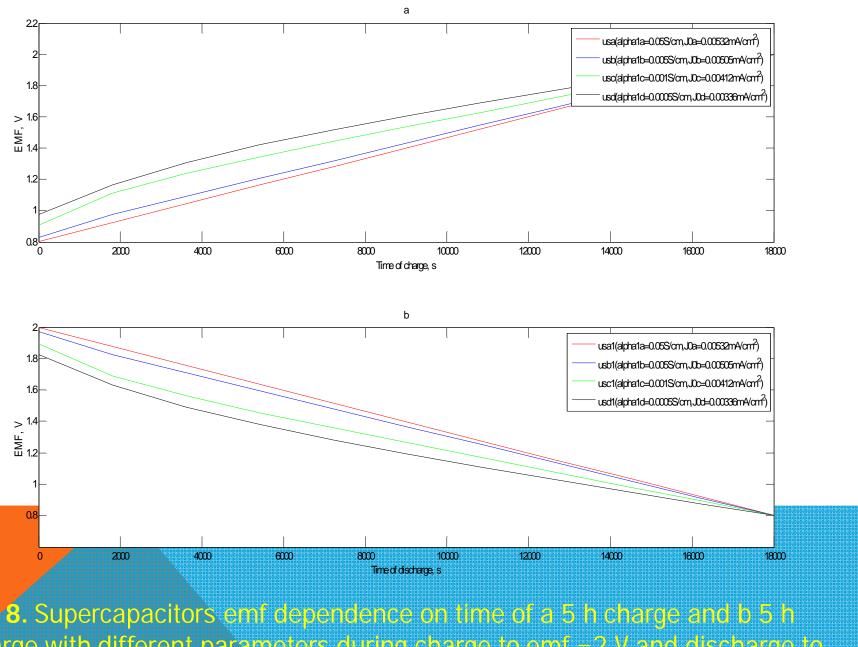


Figure 8. Supercapacitors emi depende on time of a 5 h charge and b 6 h discharge with different parameters during charge to emf = 2 V and discharge to

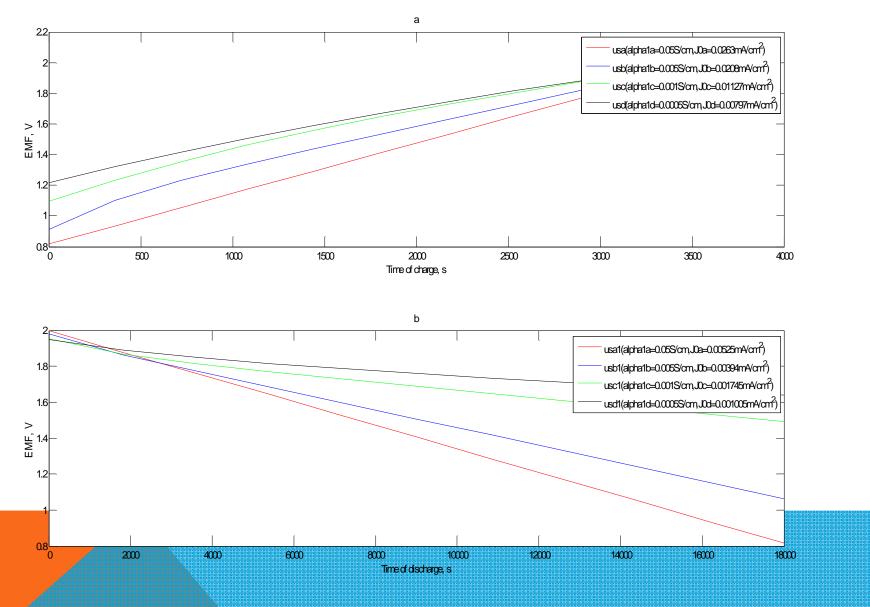


Figure 9. Capacitor emf dependences on the time of (a) 1 h charge and (b) 5 h discharge with different parameters during charge emf = 2 V and discharge to emf = 0.8 V.

Table 1. Parameters of the Supercapacitors during charge and discharge in mode I and II

		(S/cm) Alpha1 = Alpha2 = Alpha								
			Mode I				Mode II			
S/N	parameter	units	0.0005	0.001	0.005	0.05	0.0005	0.001	0.005	0.05
1	$I_{ch}$	Α	21.134	25.915	31.76	33.46	50.13	70.89	130.83	165.427
2	$Q_{\ ch}$	Ah	105.67	129.574	158.82	167.31	50.13	70.89	130.83	165.427
3	$E_{Sch}$ F eff	Wh	165.99	196.83	227.32	234.576	80.258	113.26	198.2	233.12
4	$L_{c_{-l}}$	Wh	127.22	164.736	217.24	233.989	56.617	80.258	166.793	230.237
5	$F^{Rpot}$	Wh	38.77	32.094	10.08	0.587	23.641	32.463	31.407	2.883
6	$U_{Sch}^{r}$	V	1.556	1.727	1.936	1.997	1.159	1.307	1.736	1.9835
7	$E_{Sch}^{ap}$	Wh	124.48	163.71	217.22	233.987	49.15	74.66	165.89	230.23
8	$E_{ch}^{dpol}$	Wh	2.74	1.026	0.02	0.002	7.467	6.137	0.903	0.007
9	$E_{ch}^{R}$	Wh	4.26	6.406	9.625	10.68	4.7945	9.587	32.655	52.208
10	$E_{ch}$	Wh	170.25	203.24	236.94	24526	85.053	122.847	230.855	285.328
11	$I_{dis}$	Α	13.33	20.065	30.066	33.4	6.32	10.976	24.78	30.02
12	$Q_{dis}$	Ah	66.67	100.3	150.33	167.0	31.6	54.88	123.91	165.11
13	$E_{Sdis}$	Wh	71.319	117.40	201.21	233.15	29.346	55.014	154.082	229.486
14	$E_{\mathbf{r}_{adis}}$	Wh	37.728	27.07	7.068	0.252	16.337	13.886	5.7386	0.3082
15	$oldsymbol{L}_{dig}$	Wh	15.433	19.24	8.942	0.585	3.467	5.760	6.0694	0.4358
16	$U_{\rm Sdis}^{ap}$	V	1.079	1.0094	0.8607	0.8022	0.9325	0.9145	0.8495	0.80226
17	$E_{dis}$	Wh	36.64	26.48	7.045	0.246	16.04	13.72	5.706	0.254
18	$E_{dis}^{dpol}$	Wh	1.087	0.59	0.023	0.006	0.297	0.166	0.0326	0.0544
19	$E_{dis}^{R}$	Wh	1.696	3.84	8.479	10.64	0.381	1.149	5.857	8.596
20	$E_{dis}$	Wh	69.62	113.56	192.73	222.51	28.965	53.865	148.225	220.89
21	<b>∠</b> Sadis	Ah	39.0	29.27	8.49	0.31	18.53	16.01	6.92	0.317
22	$\eta_{E1}$	%	40.89	55.87	81.34	90.72	34.055	43.847	64.207	77.416
23	$\eta_{E2}$	%	62.41	68.9	84.315	90.71	52.914	55.024	66.678	77.505
24	$O_{ERpol}$	%	31.83	25.24	7.9	0.477	31.87	31.114	16.246	1.163
25	$\delta_{\it ER}  \delta_{\it Ednol}$	%	3.498	5.04	7.51	8.6928	6.085	8.740	16.683	21.31
26	$O_{Ednol}$	%	2.247	0.795	0.422	0.003	9.128	5.126	0.405	0.02

## CONCLUSIONS

The energy, capacity, power parameters, energy efficiency of charge—discharge cycles of the capacitors will depend on:

- type and value of the effective conductivity of the solid matrixes of electrodes with DEL
- the effective conductivity of electrolyte
- thickness and specific capacitance of electrodes
- values of charge and discharge currents of the supercapacitors
- and the time of charge and discharge of supercapacitors.

### Acknowledgements









DST/NRF Centre of Excellence In Strong Materials, COE-SM, School of Chemical and Metallurgical Engineering, and

Materials for Energy Research Group, MERG, University of the Witwatersrand, Johannesburg, South Africa.

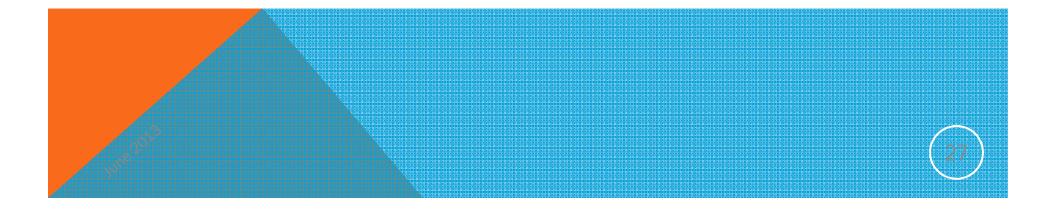
Prof. lakovos Sigalas

Prof. Sunny E. lyuke

Prof. Kenneih I. Ozoemena



# Thank you for your attention



### **COMMENTS**

### CONTRIBUTIONS

# Questions



Mathematical modelling, design, and optimization of electrochemical capacitors from layered materials Ike, I. S.; Iakovos, S.; Iyuke, S. and Ozoemena, K. I. is licensed under a <u>Creative Commons Attribution-NonCommercial-NoDerivatives 4.0 International License</u>